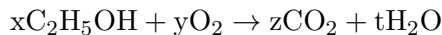


## Gauss elimination and chemistry.

**Problem:** balance the chemical reaction



**Solution:** since the number of atoms of each kind before and after the reaction is the same, we obtain the following homogeneous system of linear equations

$$\begin{cases} 2x - z = 0 \\ 6x - 2t = 0 \\ x + 2y - 2z - t = 0 \end{cases}$$

The system is solved by Gauss method as follows:

$$\begin{aligned} & \left( \begin{array}{cccc|c} 2 & 0 & -1 & 0 & 0 \\ 6 & 0 & 0 & -2 & 0 \\ 1 & 2 & -2 & -1 & 0 \end{array} \right) \xrightarrow{R_{13}} \left( \begin{array}{cccc|c} 1 & 2 & -2 & -1 & 0 \\ 6 & 0 & 0 & -2 & 0 \\ 2 & 0 & -1 & 0 & 0 \end{array} \right) \xrightarrow{R_2-6R_1; R_3-2R_1} \\ & \left( \begin{array}{cccc|c} 1 & 2 & -2 & -1 & 0 \\ 0 & -12 & 12 & 4 & 0 \\ 0 & -4 & 3 & 2 & 0 \end{array} \right) \xrightarrow{-\frac{1}{12}R_2} \left( \begin{array}{cccc|c} 1 & 2 & -2 & -1 & 0 \\ 0 & 1 & -1 & -\frac{1}{3} & 0 \\ 0 & -4 & 3 & 2 & 0 \end{array} \right) \xrightarrow{R_3+4R_2} \\ & \left( \begin{array}{cccc|c} 1 & 2 & -2 & -1 & 0 \\ 0 & 1 & -1 & -\frac{1}{3} & 0 \\ 0 & 0 & -1 & \frac{2}{3} & 0 \end{array} \right) \xrightarrow{-R_3} \left( \begin{array}{cccc|c} 1 & 2 & -2 & -1 & 0 \\ 0 & 1 & -1 & -\frac{1}{3} & 0 \\ 0 & 0 & 1 & -\frac{2}{3} & 0 \end{array} \right) \xrightarrow{R_2+R_3; R_1+2R_3} \\ & \left( \begin{array}{cccc|c} 1 & 2 & 0 & -\frac{7}{3} & 0 \\ 0 & 1 & 0 & -1 & 0 \\ 0 & 0 & 1 & -\frac{2}{3} & 0 \end{array} \right) \xrightarrow{R_1-2R_2} \left( \begin{array}{cccc|c} 1 & 0 & 0 & -\frac{1}{3} & 0 \\ 0 & 1 & 0 & -1 & 0 \\ 0 & 0 & 1 & -\frac{2}{3} & 0 \end{array} \right) \end{aligned}$$

The last matrix is of reduced row echelon form, and we obtain that the solution of the system of linear equations is  $t = s$ ,  $z = \frac{2}{3}s$ ,  $y = s$ ,  $x = \frac{1}{3}s$ , where  $s$  is a parameter. Therefore the minimal solution with positive integral  $x, y, z$  and  $t$  is obtained for  $s = 3$ :  $x = 1, y = 3, z = 2, t = 3$ . Thus, the balanced reaction is

